ne: _					Food Scien
ight in cor sol	ove n Far nstru ve p	the mily ict n	heal and nean lems	th and quality of life of Missouri citizens, performance co d Consumer Sciences Education programs enable students ing related to the biochemistry of foods and nutrition; related to foods and nutrition through the application of act of food safety and sanitation on the health and well-be	s to: scientific principles; and,
	e the			by checking the appropriate number or letter to indicate lect employability readiness rather than the grades given	
ting					
				can work independently with no supervision upervision – can perform job completely with limited supervision –	pervision
				red – requires instruction and close supervision	VISION
N	No 1	Exp	osu	re – no experience or knowledge in this area	
3	2	1	N	A. Orientation to Food Science	Notes:
				1. Compare and contrast food science to foodservice management	
				Utilize basic principles of measurement in scientific	
				experimentation (e.g., metrics, formulas, and	
				equations)	
				3. Demonstrate use, care, and safety of scientific lab	
				equipment	
				4. Explain the steps in the scientific method	
				5. Write accurate and complete reports of science	
				experiments	
				6. Identify the public and private organizations that	
				influence food service, dietetics and nutrition	
				industries 7. Identify career paths related to food science,	
				dietetics and nutrition	
				Other:	
3	2	1	N	B. Sensory Evaluation of Food	Notes:
				1. Identify qualities that make up the sensory	
				characteristics of food	
				2. Describe characteristics of sensory evaluation using	
				appropriate terms 3. Determine characteristics that affect food	
				preferences	
-				Other:	

3	2	1	N	C. Food Safety and Sanitation	Notes:
				1. Compare the positive and negative effects of yeast,	
				molds, bacteria and enzymes in foods	
				2. Identify principles of HACCP (assess hazards,	
				identify critical control points, set up control	
				procedures, monitor critical control points, take	
				corrective actions, develop a record keeping system,	
				verify that the system is working)	

				3. Describe the types of microorganisms that cause	
				foodborne illness (e.g., bacteria, viruses, parasites,	
				yeast, molds)	
				4. Explain the relationship between microorganisms	
				and foodborne illness	
				5. Describe the basic environmental conditions that	
				encourage the growth of microorganisms (e.g., time,	
				temperature, moisture, oxygen)	
				6. Identify the three major types of hazards that cause	
				foodborne illness (biological, chemical and physical)	
				7. Describe symptoms and causative agents of major	
				foodborne illnesses (e.g., salmonellosis, botulism,	
				hepatitis A)	
				8. Discuss how contamination and cross-	
				contamination of foods can occur	
				9. Identify methods and procedures for controlling	
				foodborne illness	
				10. Demonstrate personal hygiene/health practices	
				essential for food safety and sanitation	
				11. Name typical products, tools and methods for	
				effective cleaning and sanitizing	
				12. Identify location of and information on MSDS	
				(Material Safety Data Sheets)	
				Other:	
					<u> </u>
2	2	1	N ⊺	D. Sajanga Faundations	Notes:
J	2	1	N	D. Science Foundations Demonstrate the effects of acids and bases in food	ivotes:
				metabolism, preparation, processing and preservation	
				2. Describe the classes of matter, including pure	
				substances and mixtures	
				3. Identify the elements found in biochemical systems	
				(food) and their atomic symbols	
				4. Explain the properties and principles of matter and	
				energy (e.g., bonding, parts of the atom) 5. Identify components of a chemical equation	
				3. Identity components of a chemical equation	
				1	
				6 Differentiate hetween chamical reactions and	
				6. Differentiate between chemical reactions and	
				physical changes in food	
				physical changes in food 7. Demonstrate the relationship between energy,	
				physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions	
				physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion	
				physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion and temperature	
				physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion	
				physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion and temperature 9. Explain how heat is transferred	
				physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion and temperature 9. Explain how heat is transferred 10. Interrelate the effects of temperature, latent heat	
				physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion and temperature 9. Explain how heat is transferred 10. Interrelate the effects of temperature, latent heat and phase changes	
				physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion and temperature 9. Explain how heat is transferred 10. Interrelate the effects of temperature, latent heat	
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3			N	physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion and temperature 9. Explain how heat is transferred 10. Interrelate the effects of temperature, latent heat and phase changes Other:	
3	2	1	N	physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion and temperature 9. Explain how heat is transferred 10. Interrelate the effects of temperature, latent heat and phase changes Other: E. Biochemistry of Foods and the Major Nutrient	
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3	2	1	N	physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion and temperature 9. Explain how heat is transferred 10. Interrelate the effects of temperature, latent heat and phase changes Other: E. Biochemistry of Foods and the Major Nutrient Groups (Carbohydrates, Proteins, Lipids, Vitamins, Minerals and Water) 1. Describe terms related to the major nutrients and nutrient groups	
3	2	1	N	physical changes in food 7. Demonstrate the relationship between energy, physical changes, and chemical reactions 8. Discuss the relationship between molecular motion and temperature 9. Explain how heat is transferred 10. Interrelate the effects of temperature, latent heat and phase changes Other: E. Biochemistry of Foods and the Major Nutrient Groups (Carbohydrates, Proteins, Lipids, Vitamins, Minerals and Water) 1. Describe terms related to the major nutrients and nutrient groups 2. Describe the functions of the major nutrients and	
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4. Explain the process of sugar hydrolysis	
5. Identify the product resulting from the hydrolysis of	
sucrose and lactose	
6. Compare the structures of amylose and amylopectin	
and their effect on cooking properties	
7. Discuss gelatinization, paste, retrogradation and	
syneresis as they relate to starch cookery	
8. Explain what occurs during denaturization	
9. Describe the chemical structure of protein	
10. Compare the properties of saturated and unsaturated fatty acids	
11. Identify foods that contain saturated and	
unsaturated fats (triglycerides)	
12. Explain the differences between the types of fats	
and cholesterol	
13. List the ways lipid oxidation can be controlled in	
food	
14. Explain the role of water in the formation of	
solutions, colloidal dispersions and emulsions	
15. Compare and contrast bound and free water in	
foods	
16. Compare and contrast water soluble and fat soluble vitamins	
17. Distinguish between major and trace minerals	
18. Identify some interrelationships among nutrients	
(e.g., vitamin D and calcium)	
19. Discuss the effect of food processing on vitamin	
and mineral levels in food	
20. Relate metabolism to the factors that affect it	
Other:	

3	2	1	N	F. Chemistry of Food Formulations and Reactions	Notes:
				1. Explain the relationships between enzymes,	
				coenzymes and substrates	
				2. Identify factors that affect enzymatic activity	
				3. Distinguish the function of enzymatic reactions in food spoilage and food preparation	
				4. Identify the solvents and solutes in solutions	
				5. Analyze the effect of concentration on physical properties of a solution	
				6. Identify the properties of colloidal dispersions	
				7. Explain the relationship of an emulsion's parts	
				8. Explain the functions and properties of leavening agents	
				9. Identify the classes of food additives	
				10. Discuss the role of governmental regulations regarding food additives	

		11. Explain the use of additives in food	
		12. Discuss the risks and benefits of using additives in food	
		Other:	

3	2	1	N	G. Food Manufacturing Processes (Fermentation,	Notes:
				Canning, Freezing, Dehydration and Irradiation)	
				1. Define terms related to food manufacturing	
				processes	
				2. Identify commonly processed foods	
				3. Compare and contrast food manufacturing	
				processes	
				4. Identify equipment used in food manufacturing	
				processes	
				Other:	